

Corrosion behaviour of a self-pierce rivet joint of carbon-fibre reinforced plastic-laminate and EN AW-6060-T6 subjected to laboratory and outdoor exposure tests



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Outline

- Light-weight design concept
- Joining by riveting
- Corrosion
- Experimental
 - Indoor and Outdoor exposure
- Results
- Conclusion



AUDI A8 SPACE FRAME – MATERIALIEN

Vakuum Strukturguss (Alu)	35 %
Blech (Alu)	35 %
Extrusionsprofile (Alu)	22 %
Hochfestes Blech (Stahl)	8 %



Source: Audi

Light-weight design concept

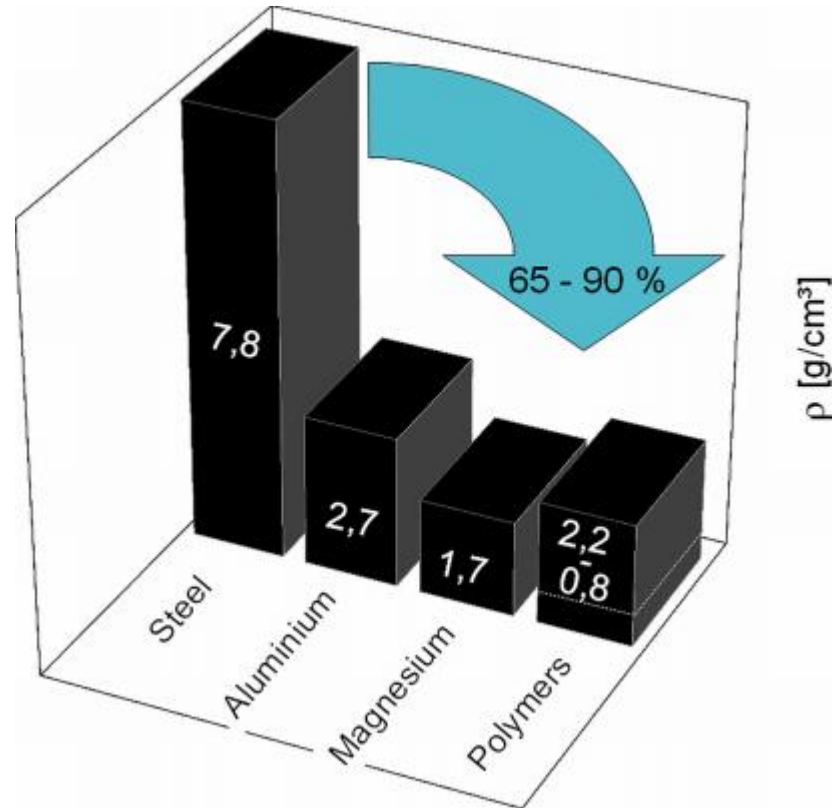
Weight reduction → replacement of steel → CO₂ reduction

Alternatives:

Aluminium

Magnesium

Polymers



→ Good atmospheric corrosion behaviour

→ Multi-Material design needs appropriate joining technologies

Corrosion

Atmospheric corrosion is influenced by:

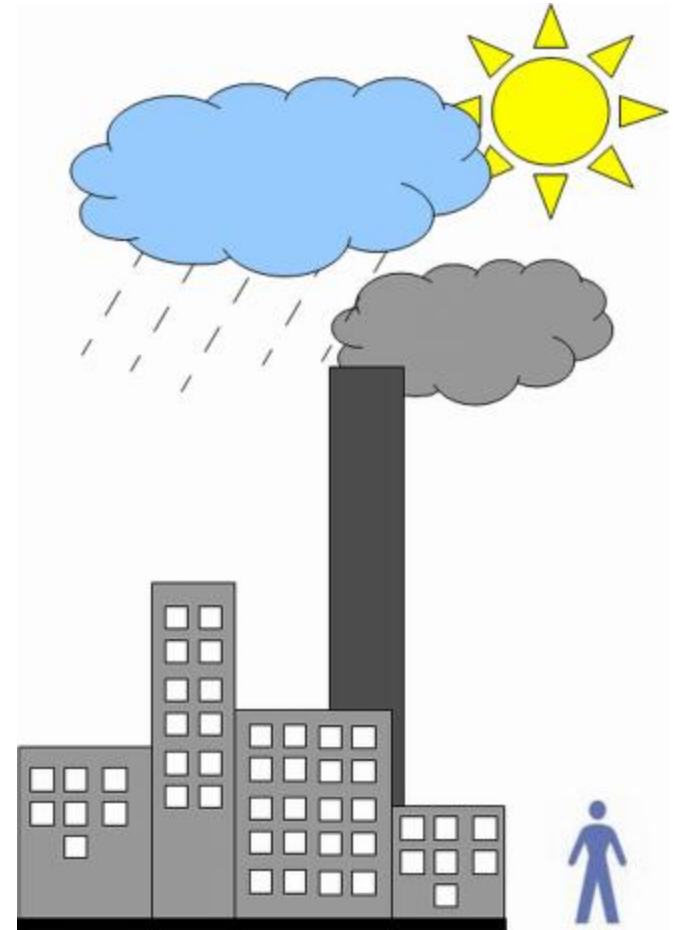
- Relative humidity
- Fly-ash
- Temperature
- Gases
- Solved salts
- Human factors

...

Good behaviour of intrinsic corrosion

but problems when joined

→ Galvanic Corrosion, Crevice Corrosion



Corrosive effects

Joining by riveting



Source: Wilhelm Böllhoff GmbH & Co. KG

Advantages

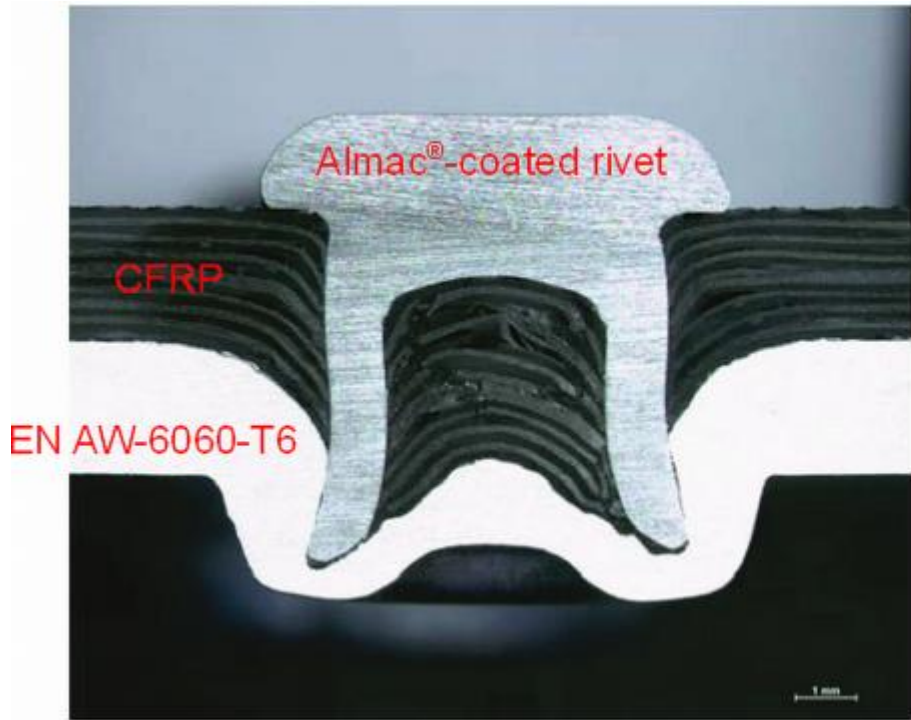
- No heat introduction
- No pre-drilled holes
- High dynamic tensile strength
- Force- and form-fit
- Appropriate for hybrid-compounds

Disadvantages

- Access to both sides
- High force load
- Galvanic corrosion

DFG-AiF Cluster Project KOMMA

- Investigation of 2 rivet joints in
 - Interaction of corrosion and mechanical load



Self-piercing rivet joint
CFRP/Al

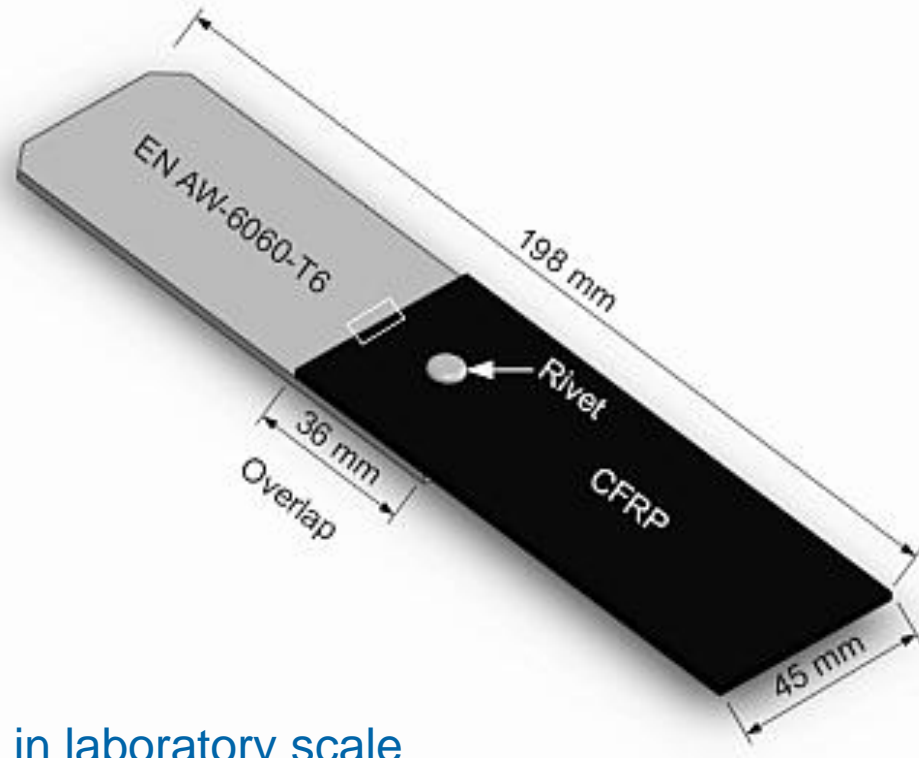


Blind rivet joint
Steel/Al

Self-piercing rivet joint

- Extruded sheet of EN AW-6060-T6
- CFRP-laminate (T700SC/RIM935)
- Almac[®]-coated self-pierce rivet

Electric conductivity between EN AW-6060-T6 and CFRP-laminate due to joining process

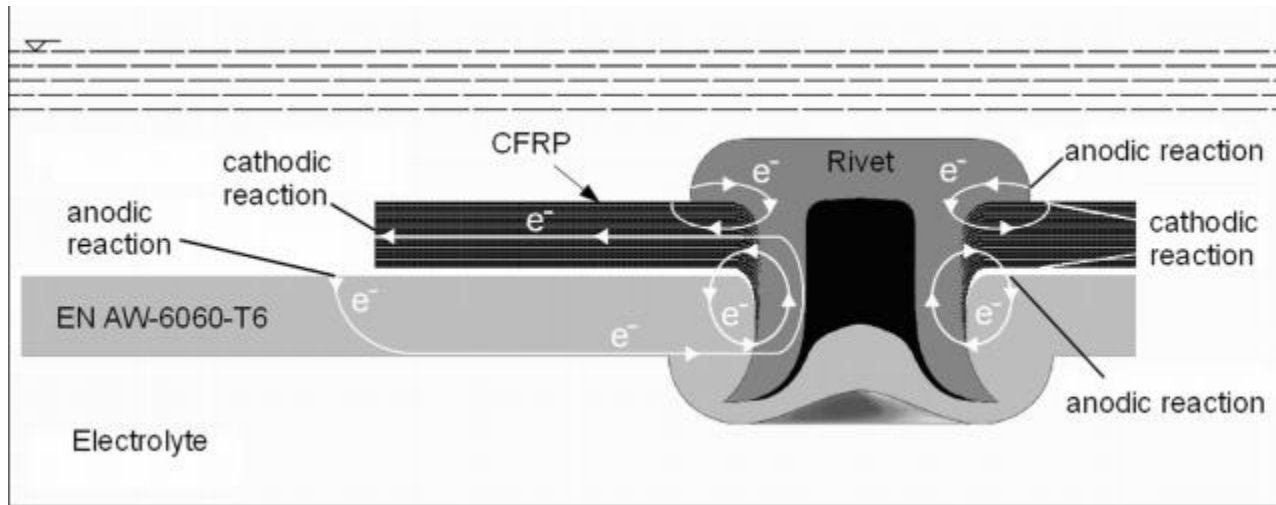


Investigation of corrosion mechanisms in laboratory scale

→ But how to transfer the results to real corrosion behaviour?

Galvanic Corrosion

- in contact to an electrolyte a galvanic element is formed
- CFRP is the cathode
- EN AW-6060-T6 and rivet are the anodes



Test conditions

Indoor



Climatic conditions defined
by controller

Outdoor



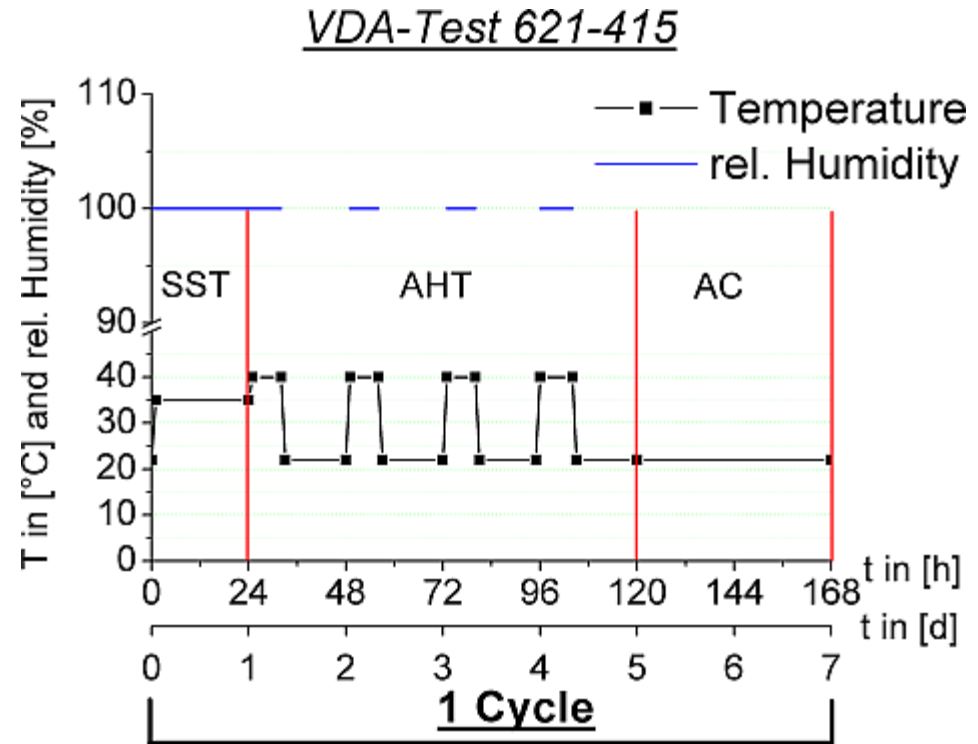
Source: Institute for Corrosion Protection Dresden GmbH

Climatic conditions defined
by environment and controller

Indoor – VDA 621-415

- 24h salt spray test (SST)
- 96h alternating humidity and air temperature test (AHT)
- 48h ambient climate test (AC)

Preferred test for investigation of subsurface corrosion of organically-coated materials



Outdoor – VDA 621-414

- Climatic conditions of Dresden
- Corrosion category C2/C3
- Weekly sprayed with 3wt% sodium chloride solution



Source: www.wetterdienst.de



Source: Institute for Corrosion Protection Dresden GmbH



Indoor – VDA 621-415

Cycle 1

- white rust at rivet head
- heavy formation of corrosion products at the overlap

Cycle 4

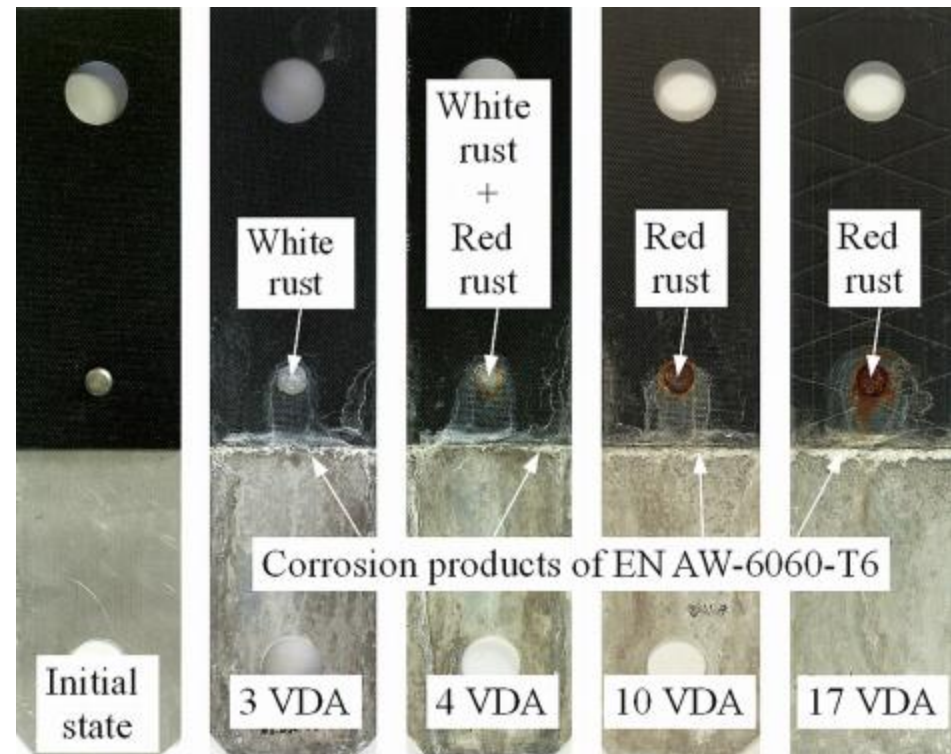
- white and red rust at rivet head

Cycle 10

- completely dissolved Almac[®]-coating

Cycle 17

- increased formation of corrosion products at the overlap

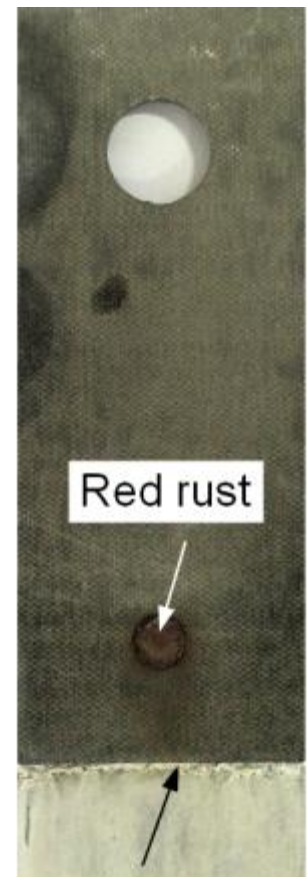


Outdoor – VDA 621-414

- red rust at rivet head
- heavy formation of corrosion products at the overlap

→ similar corrosion behaviour in comparison to indoor testing

⇒ VDA 621-415 is comparable to VDA 621-414



Corrosion products of EN AW-6060-T6



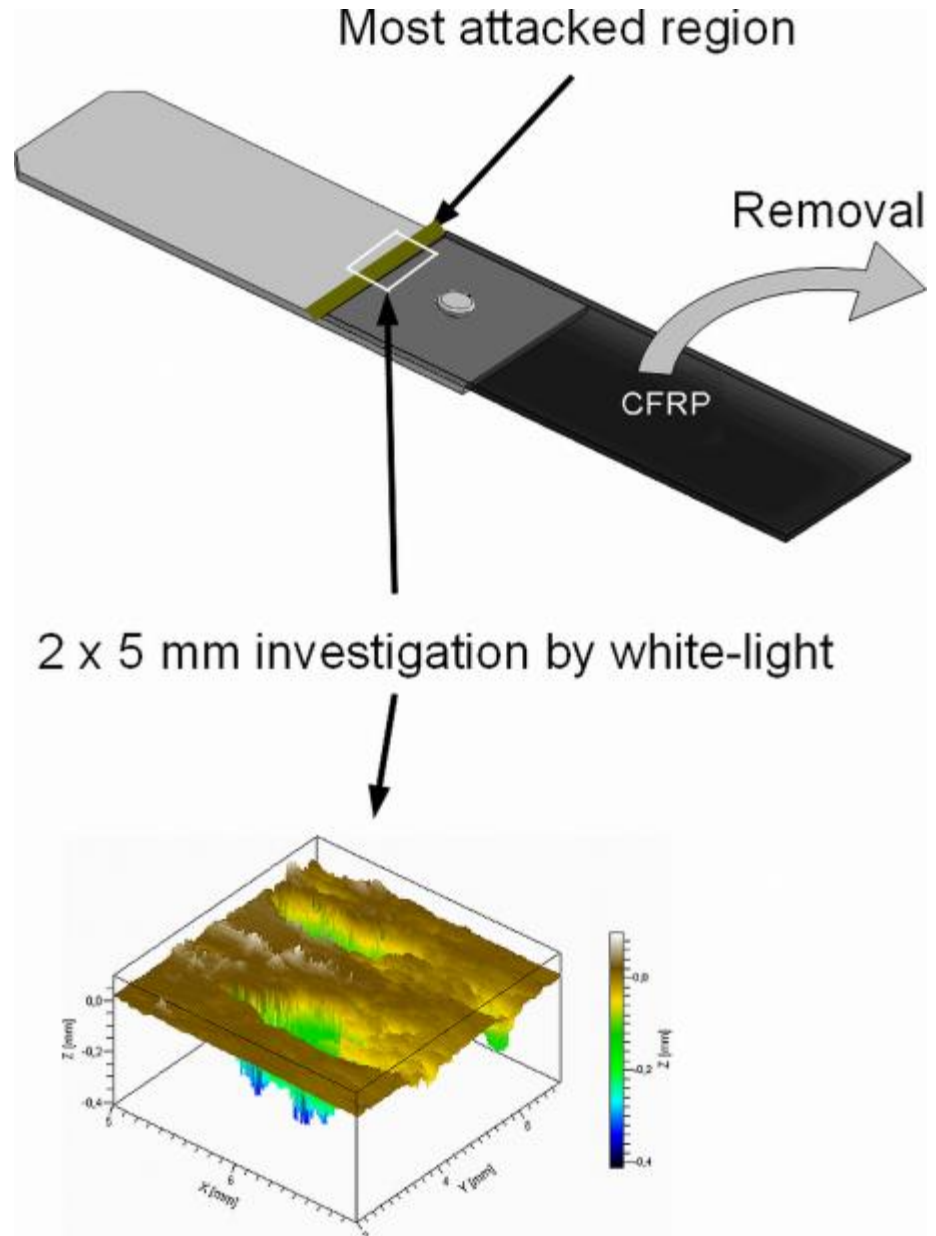
Results

Surface scan

- removal of CFRP
- most attacked region at the EN AW-6060-T6 at the overlap
- investigation by white-light interferometry

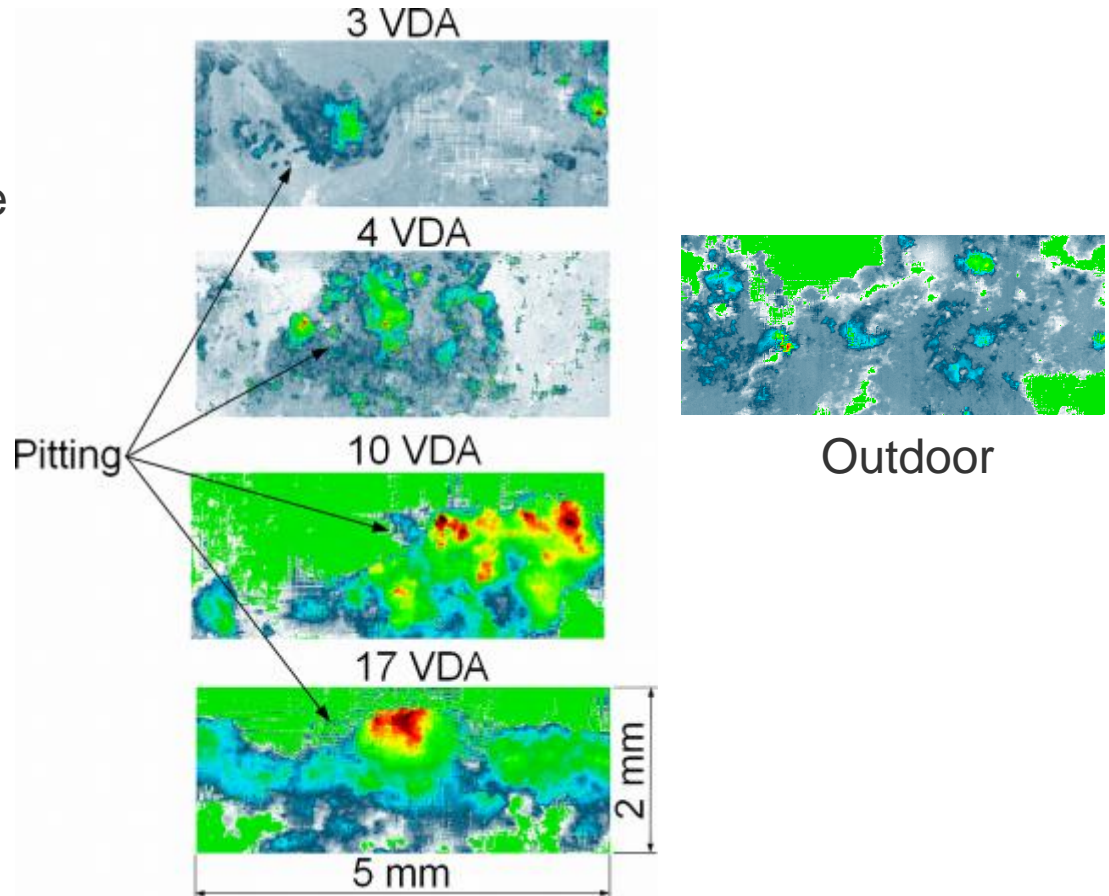
Indoor: every cycle

Outdoor: after one year



Surface scan

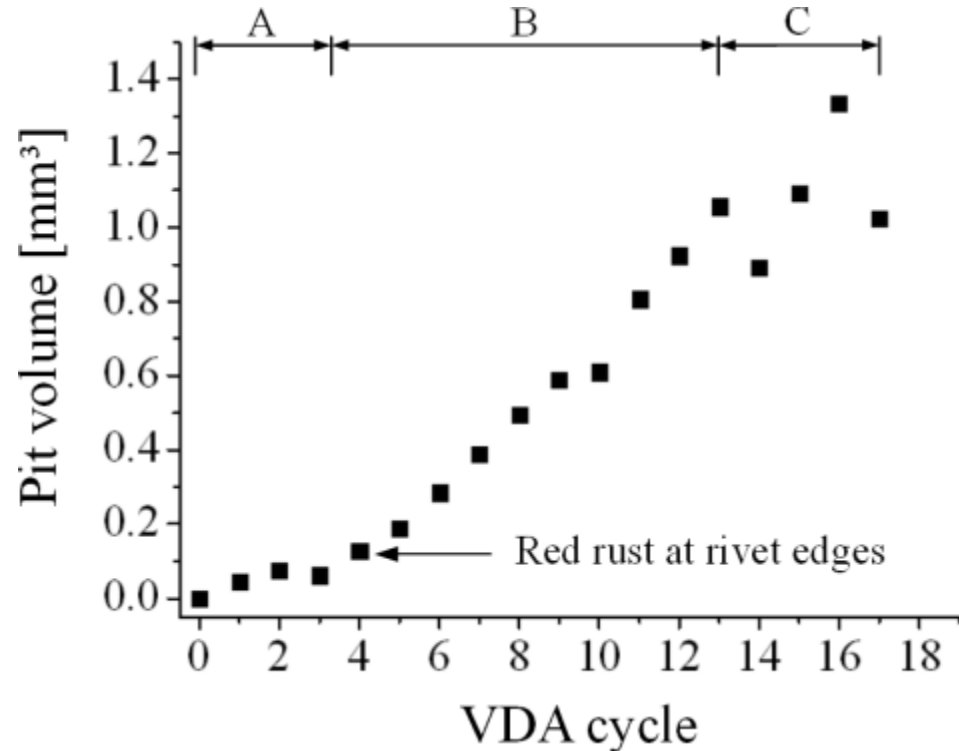
- Similar corrosion damage
- 1 year outdoor is more aggressive than 4 cycles indoor



Indoor

Pit volume

- Determination of the pit volume after each cycle of VDA 621-415
- Dividing in Region A, B and C
- complete dissolution of the Almac[®] coating



(mean of 3 samples)

Indoor – VDA 621-415

Region A: corrosion is low

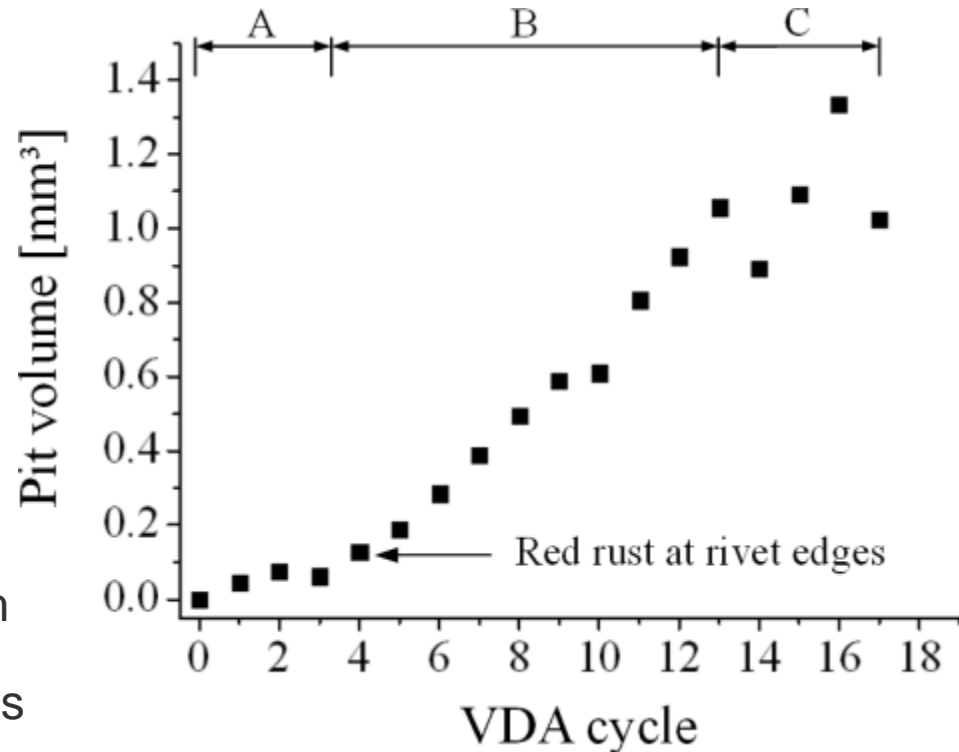
→ Pitting corrosion,
 $z \sim 0.15\mu\text{m}$

Region B: corrosion is high

→ Pitting corrosion,
 $z \sim 0.2\mu\text{m} \rightarrow z \sim 0.5\mu\text{m}$

Region C: corrosion rate decreases

→ $z \sim 0.6\mu\text{m}$

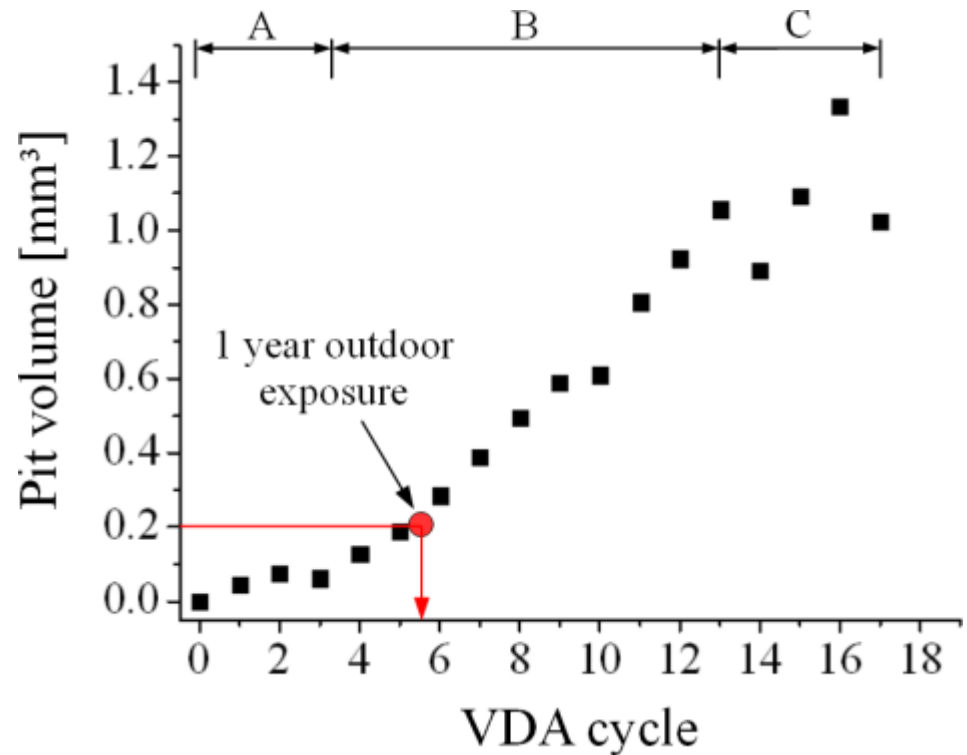


Corrosion attack as well in lateral direction → Pits coalesce

Comparison of indoor and outdoor

Corrosion attack was more aggressive than 4 cycles indoor, $z \sim 0.25\mu\text{m}$

Pit volume is equivalent to 5 – 6 cycles indoor exposure



1 year outdoor = 5 - 6 cycles indoor

Results

Similar corrosion behaviour at indoor and outdoor testing

→ Test results are comparable to each other

→ 1 cycle VDA 621-415 = 1 week

→ 1 year outdoor exposure = 5 – 6 VDA 621-415

⇒ Corrosion behaviour at 1 year outdoor exposure is received in

5 – 6 weeks indoor exposure

⇒ significant reduction of test duration

$$\text{Test-duration: Indoor} = \frac{1}{9} \text{ Outdoor}$$

⇒ Time saving of 90%

Conclusion

- Investigation of the corrosion behaviour of a self-piercing rivet joint

EN AW-6060-T6 / CFRP-laminate / Almac[®]-coated rivet

- Rivet supplies electric conduction between

EN AW-6060-T6 and CFRP

→ Formation of a galvanic element in the presence of an electrolyte

Anodes: EN AW-6060-T6, Almac[®]-coated rivet

Cathode: CFRP-laminate



Conclusion

- Investigation of the corrosion behaviour by 17 cycles of VDA 621-415 and 1 year VDA 621-414
 - Corrosion damage is similar
 - Test results are comparable to each other
 - Determination of the pit volume in the most attacked region of EN AW-6060-T6
 - 1 year VDA 621-414 = 5 – 6 cycles of VDA 621-415
- ⇒ Significant reduction of test duration